

# Electrochemical Production of Long Chain Hydrocarbons and Oxygenates

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In this presentation, we share our works related to the development of catalysts and discovery of the mechanisms for the electroreduction of CO<sub>2</sub> and CO to long-chain oxygenates, such as 1-butanol and C<sub>3</sub>-C<sub>6</sub> acetate esters, and long-chain hydrocarbons. In particular, we shall show how C<sub>1</sub>-C<sub>6</sub> alkanes and alkenes such as *n*-hexane could be selectively formed. The catalysts that were found to work well were unexpectedly based on nickel and cobalt, rather than copper (which is generally known to reduce CO<sub>2</sub> to C<sub>1</sub>-C<sub>3</sub> molecules).

1,3-butadiene is industrially produced as a by-product alongside ethylene from the energy-intensive thermal cracking of naphtha. We shall show that copper catalysts, after a modification with iodide anions, are efficacious for electro-converting acetylene to 1,3-butadiene at ambient temperature and pressure. 1,3-butadiene could be produced with a Faradaic efficiency of 93% at -0.85 V versus the Standard Hydrogen Electrode (SHE) and a partial current density of -75 mA cm<sup>-2</sup> at -1.0 V versus SHE. The partial current density of 1,3-butadiene was at least 20 times higher than that reported in previous works. Characterisation of the catalyst using in situ spectroscopies and density functional theory calculations revealed that iodide promotes stable ensembles of neutral and partially oxidised Cu sites (Cu<sup>δ+</sup>-Cu<sup>0</sup> sites), which enhance the C-C coupling of \*C<sub>2</sub>H<sub>3</sub> intermediates to form 1,3-butadiene. We shall also showcase strategies to couple acetylene into longer-chain hydrocarbons, which could potentially be used as aviation fuel.

## References

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